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⑤④ Surface treatment method of heat-resistant alloy.

⑤⑦ A method of surface treatment of a member made of heat-resistant alloy comprises spraying onto the surface of said member as a first layer a coating of a heat resistant material comprising for example a metal such as Ni or Cr or a Ni-Cr alloy or a compound thereof. A liquid coating containing a corrosion resistant material is then applied as a second layer on to the first layer. The member is then heat treated to effect penetration by diffusion of one coating into the other.

**EP 0 048 083 A1**

-1-

"Surface Treatment Method of Heat-Resistant Alloy"

This invention relates to a method of surface treatment of a member of heat-resistant alloy for use in turbines, blowers, boilers or the like to render it resistant to high temperature oxidation as well as to high temperature corrosion.

In industrial gas turbines using petroleum or natural gas as the fuel, gas temperature at the turbine inlet tends to become higher as the turbine efficiency is improved. On the other hand, as the available fuel supply has changed for the worse in recent years, the fuels used for the turbines have been diversified and the content of corrosive impurities in the fuels such as sulphur (S), sodium (Na), vanadium (V), and so forth has tended to increase. As a result, so-called "hot parts" such as the blades and burners of turbines, that are exposed to these high temperature gases, are subjected to extremely severe high temperature oxidation as well as high temperature corrosion.

These hot parts have conventionally been made primarily of heat-resistant alloys. In particular turbine blades consist of Ni- and Co-based alloys called "ultra-alloys". However, since high temperature strength is generally a top

-2-

priority requirement for these ultra-alloys, they have the drawback that their corrosion resistance and oxidation resistance are not satisfactory. Various attempts have therefore been made to provide these heat-resistant alloys with oxidation resistance and corrosion resistance and various surface treatment methods using for example chemical and physical techniques have been employed. However, none of these methods has been really satisfactory as regards efficiency and cost.

The present invention is directed to providing a method which overcomes the deficiencies of the previous methods. Accordingly, in order to provide a member of heat-resistant alloy with high temperature oxidation resistance and high temperature corrosion resistance, the present invention provides a surface treatment method which is characterized by the steps of coating by spraying onto the surface of said member in the form of a substrate, a heat-resistant material of metals such as Ni and Cr or Ni-Cr alloys or their compounds as a first layer, then applying, as a second layer, a liquid coating containing metals such as Al, Si, Vr, Ts and the like or their alloys or compounds as the corrosion-resistant material by means of spray-coating, brush-coating or the like, and heat-treating the coated surface.

-3-

The surface treatment method of the present invention provides the characterizing features as illustrated in Table 1 in comparison with the conventional methods.

Table 1

Method of this Invention		Conventional Methods				
Method	Spraying & Slurry coating & diffusion fusion penetration	Slurry coating & diffusion penetration	CVD* & diffusion penetration	plasma spraying	Low pressure plasma spraying	electron beam vacuum deposition
Metals	CrNi CrNiAl CrNiAlSi others	Cr Al Al-Si others	Cr Al Al-Si ..	NiCrSi NiCrAlSiY ZrO <sub>2</sub> MgO	NiCrAlY CoCrAlSiY others	NiCrAlY CoCrAlY others
Productivity	medium	great	great	medium	small	small
cast	medium	small	small	medium	great	extremely great
Utilizability	done	done	done	partly done	partly done in U.S.	partly done in U.S.
adhesion	good	fair	good	fair	good	good
corrosion resistance	good	good in low temp range, bad in high temp range	good in low temp range, bad in high temp range	considerably bad	good	good
uniformity	good	good	good	fair	fair	good
surface coarseness	good	good	bad	bad	fair	good
Overall evaluation	excellent	good	good	fair	good	good

Properties of Coating

Features of Method

\* CVD : Chemical vapor deposition

-5-

The present invention will be now described in more detail by reference to an example in accordance therewith.

A substrate of Udimet 520 (by weight 19% Cr, 12% Co, 6% Mo, 3% Ti, 2% Al, 1% Fe, Ni-Bal), widely used as an ultra-alloy for the hot parts of a gas turbine, was treated in the following sequence:

(1) After the surface of the substrate had been cleaned with an alkaline emulsion cleaning agent, steam cleaning was carried out using a Fluron type solvent. The surface was further blasted using an  $Al_2O_3$  blast.

(2) A Ni-Cr (50/50 by weight) alloy was applied as a coating to form a first layer having a thickness of about  $50\mu$  by plasma spraying.

(3) The surface of the sprayed-on first layer was blasted using  $Al_2O_3$  to remove any oxide film formed on its outermost surface.

(4) The surface of the sprayed-on first layer was coated by spraying on a coating slurry formed by dispersing Al and  $SiO_2$ , each having a particle size of about 0.1 to  $1\mu$ , in an organic carrier (alcohol, solvent naphtha, etc) to form a second layer.

(5) After these treatments, the substrate was placed in an electric furnace and was held at  $800^\circ C$ . ( $\pm 5^\circ C$ ) for

-6-

20 minutes to evaporate and remove the liquid. After being further held at  $330^{\circ}\text{C}$  ( $\pm 5^{\circ}\text{C}$ ) for 15 minutes, the substrate was withdrawn from the furnace.

(6) The substrate was held at  $1,080^{\circ}\text{C}$  for 4 hours inside a hydrogen furnace, was cooled in the furnace and was then withdrawn.

Above mentioned step (4) could be carried out using a mixture of fine Al particles with  $\text{Al}_2\text{O}_3$  powder in a mixing ratio by weight of 80/20 or 50/50 or a mixture of Al with  $\text{SiO}_2$  in a mixing ratio by weight of 80/20 or 50/50. Also step (6) could be carried out using a vacuum furnace in place of the hydrogen furnace.

Although in this example Udimet 520 has been treated by the method of the invention by way of example, similar excellent results can also be obtained when treating the surfaces of other substrates such Ni-based alloy, Co-based alloy and stainless steel.

The coated surface of the substrate provided by the above described method had an extremely smooth and flat surface and Al and Si from the second layer sufficiently penetrated by diffusion into the first layer, thereby completely eliminating the fine pores of the first layer. Hence, the composite coating was rendered wholly homogeneous.

-7-

In other words, since the melting point of Al is  $660^{\circ}\text{C}.$ , Al was fused due to the heat-treatment and penetrated into the fine pores, thus presumably rendering the surface smooth and flat. Further, it was confirmed that a part of Al and Si reached and was diffused also into the substrate.

Table 2 illustrates the results of fly-ash erosion resistance test, corrosion resistance test, and practical application test using gas turbine blades, each test being applied to a member treated by a method in accordance with the present invention and a member treated by a conventional method. The composite coating produced by the method in accordance with the present invention had a better performance in comparison with that produced by the conventional method in the fly-ash erosion resistance test and the corrosion resistance test. In the practical application test using gas turbine blades, too, the coated blade produced using the method of the present invention exhibited the tendency that the deposition amount of the fuel ash became smaller. In a thermal impact test comprising holding the testpiece at  $1,100^{\circ}\text{C}.$  for 15 minutes, then charging it into the water at  $20^{\circ}\text{C}.$  and repeating these procedures five times, the composite coating produced by the method of the present invention did not suffer peeling or cracking and had extremely good adhesion.



Table 2

Conventional method		Method of this Invention
Ni-Cr spraying (about 50 $\mu$ )		Ni-Cr spraying + slurry coating (40 $\mu$ + 30 $\mu$ )
Fly-ash erosion resistance test (fly-ash particle size 16 $\mu$ fly-ash con- centration 5g/m <sup>2</sup> ) gas flow velocity 10m/min.	• Tends to be damaged in about 5 hrs.	• No abnormality in about 5 hrs.
	• 50% of sprayed layer off in about 10 hrs.	• Tends to be damaged in about 10 hrs.
	• Sprayed layer dis- appears in about 20 hrs.	• 1/2 of coating falls off in about 20 hrs.
		• Penetration por- tion disappears in about 50 hrs.
No. of revolution of T/P 3,900 r.p.m.		
Corrosion resistance test (V <sub>2</sub> O <sub>5</sub> -Na <sub>2</sub> SO <sub>4</sub> coating, simulated combustion gas) flow; 900°C, 10 hrs.		Slight overall corrosion occurs on the surface layer portion but no abnormality occurs inside the coating.
Practical appli- cation test using gas turbine blade (Gas temp ..1,000°C Metal temp .. 800°C) 200 hrs.	• Deposition of com- bustion ash is great.	• Deposition of com- bustion ash is great.
	• 70% to 80% of sprayed layer falls off.	• Slight overall corrosion occurs but most coating remains.

\* CVD : Chemical vapor deposition

-9-

Claims.

1. A method of surface treatment of a member made of heat-resistant alloy characterised by the steps of spraying onto the surface of said member a coating of a heat-resistant material, applying a liquid coating containing a corrosion-resistant material onto the sprayed-on coating and then heat treating said member to effect penetration by diffusion of one coating into the other.
2. A method according to claim 1, characterised in that said sprayed-on coating comprises Ni or Cr or a Ni-Cr alloy or a compound of Ni and/or Cr.
3. A method according to Claim 1 or Claim 2, characterised in that said liquid coating comprises a slurry.
4. A method according to any preceding claim, characterised in that said liquid coating contains at least one of the following, Al, Si, Vr, Ts, or an alloy thereof or a compound thereof.
5. A method according to Claim 4, characterised in that the liquid coating comprises a slurry formed by dispersing Al and SiO<sub>2</sub> in a liquid carrier.
6. A method according to Claim 5, characterised in that said Al and SiO<sub>2</sub> have a particle size of about 0.1 $\mu$  to 1 $\mu$ .

-10-

7. A method according to Claim 4, characterised in that the liquid coating comprises a slurry formed by dispersing  $Al$  and  $Al_2O_3$  in a liquid carrier.
8. A method according to any preceding claim, characterised in that the heat treatment includes the step of holding the member at about  $1080^{\circ}C$  for several hours.
9. A method according to Claim 8, wherein said step in the heat treatment is preceded by a heating step to evaporate the liquid, followed by a relatively short heat treatment at about  $330^{\circ}C$ .
10. A method of surface treatment of a member made of heat resistant alloy, substantially as hereinbefore described by way of example.



European Patent  
Office

# EUROPEAN SEARCH REPORT

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DOCUMENTS CONSIDERED TO BE RELEVANT			CLASSIFICATION OF THE APPLICATION (Int. Cl.)
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	
X	<u>GB - A - 2 009 251</u> (ROLLS-ROYCE LIMITED)  * Abstract; claims; especially claims 3,6,7, 9,13-16 *  --	1-8	C 23 C 7/00 C 23 C 3/00 C 23 C 9/00 C 23 C 17/00 B 05 D 1/00
	<u>GB - A - 1 439 947</u> (UNION CARBIDE CORPORATION)  * Pages 5-10 *  --	1,2,5,7	B 05 D 1/08
	<u>US - A - 3 989 863</u> (R.P. JACKSON et al.)  * Abstract; claims *  --	1-5	TECHNICAL FIELDS SEARCHED (Int. Cl.)  C 23 C B 05 D
	<u>US - A - 3 837 894</u> (R.C. TUCKER, JR.)  * Columns 7-14 *  ----	1-5,7	
			CATEGORY OF CITED DOCUMENTS
			X: particularly relevant A: technological background O: non-written disclosure P: intermediate document T: theory or principle underlying the invention E: conflicting application D: document cited in the application L: citation for other reasons
			&: member of the same patent family. corresponding document
X	The present search report has been drawn up for all claims		
Place of search VIENNA		Date of completion of the search 30-11-1981	Examiner SLAMA